

BIOCHEMISTRY

S.NO	Submit Date	Name	Total Marks (76)	Result	1. CAMPUS	2. GENDER	3. What are chemical bonds	4. which sub atomic particles participates in chemical bonding	5. when atom are stable with absolute electron configuration of 8 valence electrons is called	6. not type of a chemical bond	7. what is the atomic number of beryllium	8. which of the following does not form anion during reaction	9. ions of period 2 and 1 elements with charges greater than +2 are unstable	10. which of the following element will give us an ionic bond	11. what are the delocalized electrons in aluminium metal	12. which one is false	13. which one is correct about electronic configuration of neon	14. what is electrovalent bond	15. an acid is Any substance that, when dissolved in water, increases the concentration of hydronium ion (H ³ O ⁺)	16. Water is amphoteric	Answer Sheet Link
1	23-02-2022	Boniface	19	25.0%	Nyeri	M	strong electrostatic forces holding atoms or ions together	protons	radicals	metallic	3	oxygen	false	chlorine and sodium	3	Electron – nucleus attractive forces	2.8.7	bond between carbon and oxygen	false	False	https://quizzory.in/answer-sheet/6215b7ff2d59bc1df6d06a9d
2	28-02-2022	Cecilia wachia	54	71.1%	Makindu	F	strong electrostatic forces holding atoms or ions together	electron	octate rule	electrostatics	3	boron	false	chlorine and sodium	3	Electron – nucleus attractive forces	2.8	Bonds that are formed by transfer of electrons from one element to the other.	true	True	https://quizzory.in/answer-sheet/621c78332d59bc1df6d41753
3	28-02-2022	James Nyoike	44	57.9%	Nairobi	M	strong electrostatic forces holding atoms or ions together	electron	radicals	metallic	4	boron	true	phosphate	3	Electron – nucleus attractive forces	2.8	Bonds that are formed by transfer of electrons from one element to the other.	true	True	https://quizzory.in/answer-sheet/621c78ededdfe81dd2387698
4	28-02-2022	samwel Gachanja	44	57.9%	siaya	M	strong electrostatic forces holding atoms or ions together	electron	duplet rule	electrostatics	6	boron	true	chlorine and sodium	2	Electron – nucleus attractive forces	2.8	Bonds that are formed by transfer of electrons from one element to the other.	true	yes	https://quizzory.in/answer-sheet/621c80489157101e846f371d
5	28-02-2022	Vincent Abuga	26	34.2%	Eldoret	M	strong electrostatic forces holding atoms or ions together	protons	octate rule	metallic	3	oxygen	true	nitrogen and hydrogen	3	Nucleus – nucleus repulsive forces	2.8	bond between chlorine and fluorine	true	Substance that poses both basic and acidic properties	https://quizzory.in/answer-sheet/621c81e6c90a0a1e3fc2eef9

S.NO	Submit Date	Name	Total Marks (76)	Result	1. CAMPUS	2. GENDER	3. What are chemical bonds	4. which sub atomic particles participates in chemical bonding	5. when atom are stable with absolute electron configuration of 8 valence electrons is called	6. not type of a chemical bond	7. what is the atomic number of beryllium	8. which of the following does not form anion during reaction	9. ions of period 2 and 1 elements with charges greater than +2 are unstable	10. which of the following element will give us an ionic bond	11. what are the delocalized electrons in aluminium metal	12. which one is false	13. which one is correct about electronic configuration of neon	14. what is electrovalent bond	15. an acid is Any substance that, when dissolved in water, increases the concentration of hydronium ion (H ₃ O ⁺)	16. Water is amphoteric	Answer Sheet Link
6	28-02-2022	Winy	55	72.4%	Kisii	F	strong electrostatic forces holding atoms or ions together	electron	radicals	electrostatics	4	boron	true	chlorine and sodium	3	Electron – nucleus attractive forces	2.8	Bonds that are formed by transfer of electrons from one element to the other.	true	True	https://quizzory.in/answer-sheet/621c824e5074781e675a67de
7	28-02-2022	Muga	39	51.3%	Kis	M	strong electrostatic forces holding atoms or ions together	electron	octate rule	metallic	4	chlorine	true	chlorine and sodium	1	Electron – protons repulsive forces	2.8.2	bond between chlorine and fluorine	true	True	https://quizzory.in/answer-sheet/621c82a0e4ff921e13aa3724
8	28-02-2022	Brian	50	65.8%	Siaya	M	strong electrostatic forces holding atoms or ions together	electron	octate rule	electrostatics	4	oxygen	true	chlorine and sodium	1	Electron – nucleus attractive forces	2.8	Bonds that are formed by transfer of electrons from one element to the other.	true	because it has both an acidic and a basic properties can react with both proper	https://quizzory.in/answer-sheet/621c84a877cfa91eb6d860f7
9	28-02-2022	Aderogba daud	63	82.9%	Udus	M	strong electrostatic forces holding atoms or ions together	electron	octate rule	electrostatics	4	boron	true	chlorine and sodium	3	Electron – protons repulsive forces	2.8	Bonds that are formed by transfer of electrons from one element to the other.	true	True	https://quizzory.in/answer-sheet/621c84a8e4ff921e13aa37e6
10	28-02-2022	Carlson Kosgey	56	73.7%	Uzima University	M	strong electrostatic forces holding atoms or ions together	electron	octate rule	electrostatics	4	sulfur	true	chlorine and sodium	3	Electron – electron repulsive forces	2.8.8	bond between carbon and oxygen	true	Substance that react in both acid and base.	https://quizzory.in/answer-sheet/621c86d854ac721e3807b744

S.NO	Submit Date	Name	Total Marks (76)	Result	1. CAMPUS	2. GENDER	3. What are chemical bonds	4. which sub atomic particles participates in chemical bonding	5. when atom are stable with absolute electron configuration of 8 valence electrons is called	6. not type of a chemical bond	7. what is the atomic number of beryllium	8. which of the following does not form anion during reaction	9. ions of period 2 and 1 elements with charges greater than +2 are unstable	10. which of the following element will give us an ionic bond	11. what are the delocalized electrons in aluminium metal	12. which one is false	13. which one is correct about electronic configuration of neon	14. what is electrovalent bond	15. an acid is Any substance that, when dissolved in water, increases the concentration of hydronium ion (H ³ O ⁺)	16. Water is amphoteric	Answer Sheet Link
11	28-02-2022	Mogire	45	59.2%	Kabarnet	M	strong electrostatic forces holding atoms or ions together	electron	radicles	electrostatistics	4	sulfur	true	chlorine and sodium	3	Electron – protons repulsive forces	2.8.2	Bonds that are formed by transfer of electrons from one element to the other.	false	Is a substance that can either react as acid or a base depending on the medium	https://quizzory.in/answer-sheet/621c8d8b54ac721e3807bc95
12	28-02-2022	Brian	62	81.6%	Nakuru	M	strong electrostatic forces holding atoms or ions together	electron	octate rule	electrostatistics	4	boron	false	chlorine and sodium	3	Electron – protons repulsive forces	2.8	Bonds that are formed by transfer of electrons from one element to the other.	true	True	https://quizzory.in/answer-sheet/621c95a43b8aa41ebc5b0b53
13	28-02-2022	Bensoul	63	82.9%	Busia	M	strong electrostatic forces holding atoms or ions together	electron	octate rule	electrostatistics	4	boron	true	chlorine and sodium	3	Electron – protons repulsive forces	2.8	Bonds that are formed by transfer of electrons from one element to the other.	true	True	https://quizzory.in/answer-sheet/621c967a54ac721e3807c265
14	28-02-2022	Jerry	51	67.1%	Bomet	M	strong electrostatic forces holding atoms or ions together	electron	octate rule	electrostatistics	4	oxygwn	true	chlorine and sodium	3	Electron – protons repulsive forces	2.8	sodium and argon	true	Yes	https://quizzory.in/answer-sheet/621c96c6e4ff921e13aa44be
15	28-02-2022	Ronoh Kipkoech	63	82.9%	Makindu	M	strong electrostatic forces holding atoms or ions together	electron	octate rule	electrostatistics	4	boron	true	chlorine and sodium	3	Electron – protons repulsive forces	2.8.2	Bonds that are formed by transfer of electrons from one element to the other.	true	because it has both acidic and basic properties.	https://quizzory.in/answer-sheet/621c9a178376901e9003ac89
16	28-02-2022	Wanjira	31	40.8%	Mosoriot	M	cohesive bond	electron	duplet rule	electrostatistics	3	oxygwn	false	chlorine and sodium	3	Electron – nucleus attractive forces	2.8	bond between carbon and oxygen	true	True	https://quizzory.in/answer-sheet/621c9d375074781e675a73cb

S.NO	Submit Date	Name	Total Marks (76)	Result	1. CAMPUS	2. GENDER	3. What are chemical bonds	4. which sub atomic particles participates in chemical bonding	5. when atom are stable with absolute electron configuration of 8 valence electrons is called	6. not type of a chemical bond	7. what is the atomic number of beryllium	8. which of the following does not form anion during reaction	9. ions of period 2 and 1 elements with charges greater than +2 are unstable	10. which of the following element will give us an ionic bond	11. what are the delocalized electrons in aluminium metal	12. which one is false	13. which one is correct about electronic configuration of neon	14. what is electrovalent bond	15. an acid is Any substance that, when dissolved in water, increases the concentration of hydronium ion (H ³ O ⁺)	16. Water is amphoteric	Answer Sheet Link
17	28-02-2022	Wilfred	45	59.2%	Siaya	M	electron forces	electron	octate rule	metallic	3	boron	true	phosphate	3	Electron – protons repulsive forces	2.8.8	Bonds that are formed by transfer of electrons from one element to the other.	true	Yes	https://quizzory.in/answer-sheet/621ca0562d59bc1df6d4332b
18	28-02-2022	Joan rono	42	55.3%	Makindu	F	strong electrostatic forces holding atoms or ions together	electron	octate rule	metallic	6	oxygwn	false	carbhorn and oxygen	3	Electron – nucleus attractive forces	2.8.8	Bonds that are formed by transfer of electrons from one element to the other.	true	Yes	https://quizzory.in/answer-sheet/621ca6183b8aa41ebc5b1423
19	28-02-2022	Kelvins Jude	56	73.7%	Kapkatet	M	strong electrostatic forces holding atoms or ions together	electron	octate rule	electrostatistics	5	oxygwn	true	chlorine and sodium	3	Electron – nucleus attractive forces	2.8.8	Bonds that are formed by transfer of electrons from one element to the other.	true	Yes,water can neither be basic or acidic	https://quizzory.in/answer-sheet/621cbb0bd3b8c51dccf08fbb
20	28-02-2022	Isaiah momanyi	69	90.8%	Kisii campus	M	strong electrostatic forces holding atoms or ions together	electron	octate rule	electrostatistics	4	boron	true	chlorine and sodium	3	Electron – protons repulsive forces	2.8.8	Bonds that are formed by transfer of electrons from one element to the other.	true	Water is amphoteric because it has both an acidic and a basic properties.can react with both bases acids	https://quizzory.in/answer-sheet/621ccae2c90a0a1e3fc3196f
21	28-02-2022	Selly chebor	30	39.5%	Iten	F	strong electrostatic forces holding atoms or ions together	electron	octate rule	electrostatistics	6	oxygwn	false	nitrogen and hydrogen	6	Nucleus – nucleus repulsive forces	2.8.2	bond between chlorine and flourine	true	Yes	https://quizzory.in/answer-sheet/621ce52b5074781e675aa4f4

S.NO	Submit Date	Name	Total Marks (76)	Result	1. CAMPUS	2. GENDER	3. What are chemical bonds	4. which sub atomic particles participates in chemical bonding	5. when atom are stable with absolute electron configuration of 8 valence electrons is called	6. not type of a chemical bond	7. what is the atomic number of beryllium	8. which of the following does not form anion during reaction	9. ions of period 2 and 1 elements with charges greater than +2 are unstable	10. which of the following element will give us an ionic bond	11. what are the delocalized electrons in aluminium metal	12. which one is false	13. which one is correct about electronic configuration of neon	14. what is electrovalent bond	15. an acid is Any substance that, when dissolved in water, increases the concentration of hydronium ion (H ³ O ⁺)	16. Water is amphoteric	Answer Sheet Link
22	28-02-2022	Faith Jeruto Rotich	32	42.1%	Lodwar	F	electron forces	neutron	octate rule	electrostatistics	4	chlorine	true	nitrogen and hydrogen	3	Electron – protons repulsive forces	2.8.7	bond between carbon and oxygen	true	A substance that contains water	https://quizzory.in/answer-sheet/621ce5bc5074781e675aa5ea
23	28-02-2022	Chepkoech Soi	68	89.5%	Sigowet	F	strong electrostatic forces holding atoms or ions together	electron	octate rule	electrostatistics	4	boron	true	chlorine and sodium	3	Electron – protons repulsive forces	2.8.8	Bonds that are formed by transfer of electrons from one element to the other.	true	False	https://quizzory.in/answer-sheet/621ce69beddfe81dd238c33c
24	28-02-2022	Chepkirui	44	57.9%	Kangundo	F	strong electrostatic forces holding atoms or ions together	protons	octate rule	electrostatistics	5	oxygwn	true	chlorine and sodium	3	Electron – protons repulsive forces	2.8	Bonds that are formed by transfer of electrons from one element to the other.	true	Elements that can react both as base and acid	https://quizzory.in/answer-sheet/621ce6e9eddf81dd238c55a
25	28-02-2022	Enock Ngeno	50	65.8%	Matibabu	M	strong electrostatic forces holding atoms or ions together	electron	radicles	electrostatistics	4	boron	false	chlorine and sodium	3	Electron – protons repulsive forces	2.8.7	bond between chlorine and flourine	true	False	https://quizzory.in/answer-sheet/621ced433b8aa41ebc5b4252
26	28-02-2022	Ric	44	57.9%	Turk	M	strong electrostatic forces holding atoms or ions together	electron	octate rule	metallic	4	oxygwn	false	chlorine and sodium	2	Electron – protons repulsive forces	2.8.8	bond between chlorine and flourine	true	Yess	https://quizzory.in/answer-sheet/621d05b95074781e675ab6eb
27	28-02-2022	Samuel ombati	49	64.5%	Makindu	M	cohesive bond	electron	octate rule	electrostatistics	4	chlorine	false	chlorine and sodium	3	Electron – nucleus attractive forces	2.8.8	Bonds that are formed by transfer of electrons from one element to the other.	false	True	https://quizzory.in/answer-sheet/621d0d6454ac721e38080d4f

S.NO	Submit Date	Name	Total Marks (76)	Result	1. CAMPUS	2. GENDER	3. What are chemical bonds	4. which sub atomic particles participates in chemical bonding	5. when atom are stable with absolute electron configuration of 8 valence electrons is called	6. not type of a chemical bond	7. what is the atomic number of beryllium	8. which of the following does not form anion during reaction	9. ions of period 2 and 1 elements with charges greater than +2 are unstable	10. which of the following element will give us an ionic bond	11. what are the delocalized electrons in aluminium metal	12. which one is false	13. which one is correct about electronic configuration of neon	14. what is electrovalent bond	15. an acid is Any substance that, when dissolved in water, increases the concentration of hydronium ion (H ³ O ⁺)	16. Water is amphoteric	Answer Sheet Link
28	28-02-2022	Alan Otieno	51	67.1%	Nakuru	M	strong electrostatic forces holding atoms or ions together	electron	octate rule	electrostatistics	4	boron	true	chlorine and sodium	3	Electron – protons repulsive forces	2.8	sodium and argon	false	No	https://quizzory.in/answer-sheet/621d0e0cd58e831e1900eb4f
29	28-02-2022	Novesturs Bett	51	67.1%	Iten	M	strong electrostatic forces holding atoms or ions together	electron	octate rule	ionic	6	boron	true	chlorine and sodium	3	Electron – protons repulsive forces	2.8	Bonds that are formed by transfer of electrons from one element to the other.	true	Aphotic has both acid and basic properties. Water therefore is not amphoteric .	https://quizzory.in/answer-sheet/621d2ef395d6691daa1237c8
30	01-03-2022	Emmanuel	56	73.7%	Chuka	M	strong electrostatic forces holding atoms or ions together	electron	octate rule	electrostatistics	3	boron	true	chlorine and sodium	3	Electron – nucleus attractive forces	2.8	Bonds that are formed by transfer of electrons from one element to the other.	true	No Answer	https://quizzory.in/answer-sheet/621e6ff8d3b8c51dccf15265